

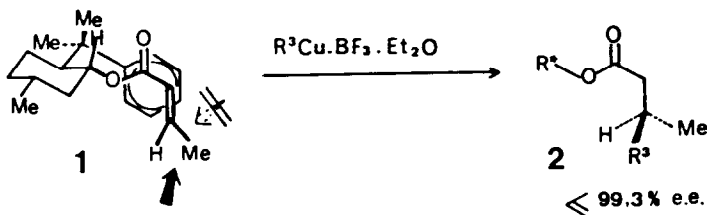
ASYMMETRIC 1,4-ADDITIONS OF COORDINATED $\text{MeCu} \cdot \text{BF}_3$ TO CHIRAL ENOATES:
ENANTIOSELECTIVE SYNTHESSES OF (S)-(-)-CITRONELLIC ACID¹

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Abstract: $n\text{Bu}_3\text{P}$ - or cyanide-stabilized $\text{RCu} \cdot \text{BF}_3$ ($\text{R} = \text{Me}$, 4-Me-3-pentenyl) undergo efficient 1,4-additions to neopentylether-shielded *trans*-enoates. Thus chiral β -substituted carboxylic acids e.g. (S)-citronellic acid were obtained in high e.e. (Schemes 2 and 4).

Recently greater than 99% enantioselective C-C bond closure β to a carboxyl group has been accomplished² by BF_3 -promoted 1,4-additions of organocopper reagents³ to the crotonate derived from (-)-8-phenylmenthol⁴ (Scheme 1). We ascribed the π -face selectivity of the reactions 1 \rightarrow 2

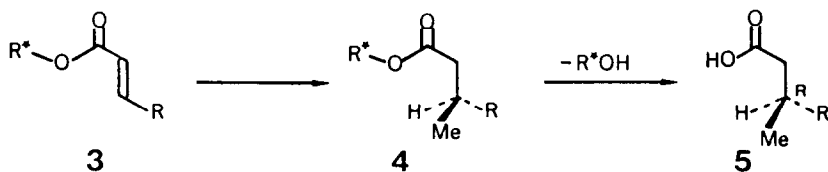
Scheme 1



to an antiplanar $\text{C}=\text{C}/\text{C}=\text{O}$ -disposition in the enoate and thus to a phenyl shielding of its C_β -*si*-face. Accordingly, by reversing the order of group introduction, either enantiomer of a β -substituted carboxylic acid should be accessible using the same chiral control element.

Prompted by the ubiquitous occurrence of methyl-substituted chiral centers in natural products, we concentrated our efforts on the conjugate addition of methylcopper derivatives to chiral enoates. Our results are depicted in the Scheme 2^{5,6}. Initial attempts to add even a large excess of $\text{MeCu} \cdot \text{BF}_3$ to the enoate 3a were impaired by the unfavorable stability/reactivity-ratio of the reagent (entry a)². By contrast, upon stabilization of the methyl-copper species by $n\text{Bu}_3\text{P}$ ¹⁶ its addition to 3a furnished the adduct 4a in 96% yield. Subsequent saponification of 4a gave the (R)-carboxylic acid 5, $\text{R} = n\text{Bu}$ in 86.5% enantiomeric excess (entry b). Realizing the increased flexibility offered by the antipodal auxiliaries 6 and 7 (Scheme 3)¹⁷ the enoate 3c was treated with various methylcopper reagents. Whereas Me_2CuLi did not add at 0° within 2 hr (entry c), Bu_3P -coordinated $\text{MeCu} \cdot \text{BF}_3$ gave the corresponding adduct 4c in 82% yield and with improved π -face differentiation (94%, entry d). A new, conveniently stable reagent obtained by mixing MeLi with CuCN (1 eq.) and $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (1 eq.)¹⁸ gave on addition to 3c followed by simple work-up the ester 4c in 80% yield with a somewhat lower induction (82% d.e. (entry e)). Comparison of entries f, d and h shows an increasing chiral induction (92% \rightarrow 98% e.e.) parallel to the size of the resident

Scheme 2



entry	R*	R	"MeCu·BF ₃ " Ligand	chem. yield % 4 ⁶	e.e.% 5 ¹⁰
a		nBu	none	28	78
b	- " -	nBu	nBu ₃ P: <u>A</u> ⁶	96	86.5
c		nBu	Me ₂ CuL1	0	—
d	- " -	nBu	nBu ₃ P: <u>B</u> ⁶	82	94
e	- " -	nBu	CN ⁻ : <u>C</u> ⁶	80	82
f	- " -	Et	nBu ₃ P: <u>B</u> ⁶	85	92
g	- " -	Et	CN ⁻ : <u>C</u> ⁶	76	80
h		nC ₈ H ₁₇	nBu ₃ P: <u>B</u> ⁶	90	98 (S)

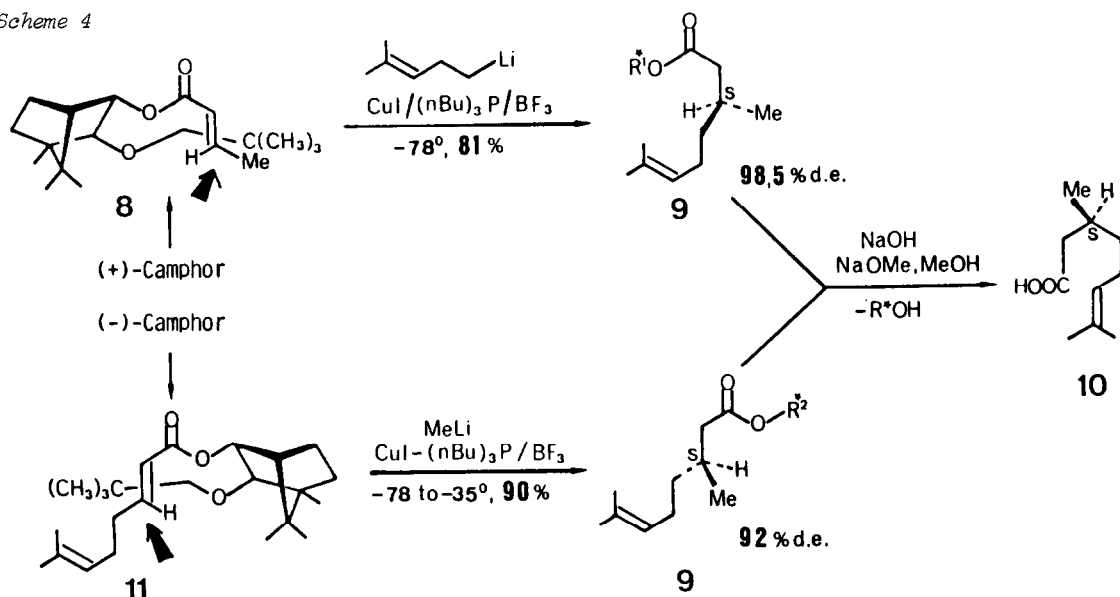
substituent R. The last entry h merits further attention since the enoate 3h derived from 7 afforded the (S)-carboxylic acid 5h with 98% overall enantioselection. Thus, alternating either the order of group introduction or the antipodal control elements, provides a high degree of flexi-

Scheme 3



bility in synthesis. This is further illustrated by the preparation of (S)-citronellic acid (Scheme 4) which is difficult to obtain in optically pure form¹⁹ and which is an interesting synthetic precursor²⁰. Addition of 4-methyl-3-pentenylcopper/BF₃/nBu₃P-complex to crotonate 8²¹ followed

Scheme 4



by saponification of the resulting product **9** afforded acid **10** efficiently with 98.5% overall enantioselectivity. Reduction of ester **9** with LiAlH_4 , gave (*S*)-citronellool (88%) together with the regenerated auxiliary **6** (81%). Permutation of both the resident group and the chirality of the control moiety is exemplified by the transformation **11** \rightarrow **9** yielding also (*S*)-citronellic acid of 92% optical purity.¹⁰

In summary we believe that asymmetric carbon-carbon bond construction by Lewis-acid promoted 1,4-additions, Diels-Alder¹⁷ and ene-reactions²² of chiral enoates are of practical value in organic synthesis. These findings are currently exploited for the syntheses of natural products. Moreover we are exploring the utility of cyanide stabilized organocopper-Lewis-acid-reagents, particularly in additions to epoxides and Michael acceptors.

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- 3 For conjugate additions of $\text{RCu}\cdot\text{BF}_3$ see: *Y. Yamamoto*, *S. Yamamoto*, *H. Yatagai*, *Y. Ishihara* and *K. Maruyama*, *J. Org. Chem.* **47**, 119 (1982).
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- ⁵ All new compounds were characterized by IR, ¹H-NMR and MS.
- ⁶ The enoates were prepared by successive treatment of the corresponding acid with (COCl)₂ and R*OH/AgCN⁷. Carboxylic acids 3c, 3h, R*=H were obtained by Horner reaction⁸. For the conversions 3 → 4 freshly prepared 1.2N solutions of MeLi in ether (MeI + Li) have been employed under argon with vigorous stirring as follows: A: CuI(10eq) add MeLi(10eq). -10°, 20 min. → -78°, add nBu₃P (10eq), 10 min, add BF₃.Et₂O(10eq), 1h, add 3 (1eq), then -10°, 6h/aq. NH₄Cl, 2 chromatographies (SiO₂); B: nBu₃P-CuI⁹ (10eq), -78°, add MeLi(10eq) over 30 min → -20° → -78°, add BF₃.Et₂O(10eq) over 30 min, 1h, add 3 (1eq) over 40 min, -78°, 2h then -35°, 16h/aq. NH₄Cl, 2 chromatographies; C: anhydr. CuCN (5eq), -78°, add MeLi (5eq) → -10° → -78° add BF₃.Et₂O (5eq), 1h, add 3 (1eq) over 45 min → -20°, 16 h/aq. sat. NH₄OH/NH₄Cl (1:22), flash chromatography. Saponification 4 → 5: 2N NaOMe in MeOH (33 eq), H₂O (5eq), 75°, 15 h (87-99%).
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- ¹⁰ The absolute configurations of 5 and 10 follow from chiroptic comparison with literature references: 5, R=nBu¹¹, 5, R=Et¹², 5h, R=nC₈H₁₇¹³, 10¹⁴. The enantiomeric purity of 5 and 10 were determined by analyses of their (R)-1-(1-naphthyl)ethylamides using HPLC¹⁵ (μ-Porasil, hexane/EtOAc 9:1) and GC (OV-1, capillary column, 170°, 0.7 kg H₂/cm²).
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- ²¹ 8 was converted to 9 using the following conditions: CuI(2eq), ether, 0°, add nBu₃P (2eq) → +20° → -65° add freshly prepared (RBr/Li) 4-methyl-3-pentenyl lithium over 30 min. → -10°, 20 min, → -78° add BF₃.Et₂O (2eq), 1.5h, add 8 (1eq) over 1h/aq. NH₄Cl, chromatography.
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